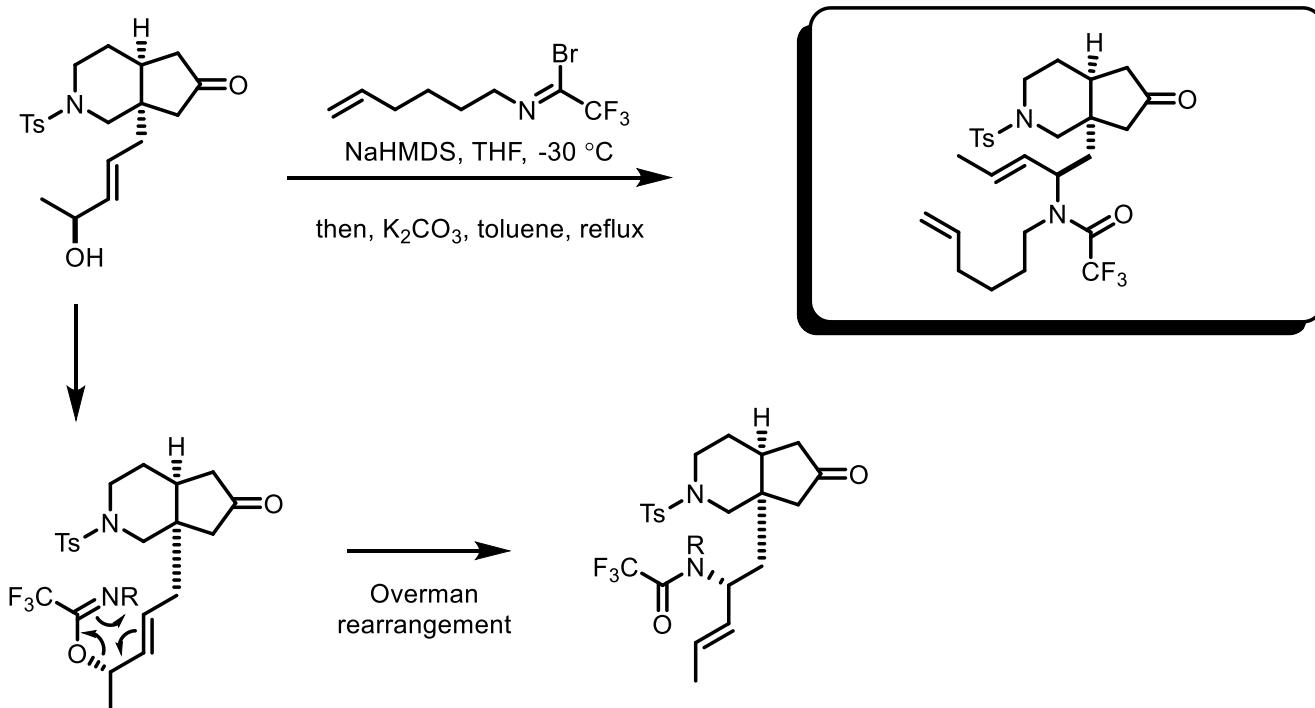
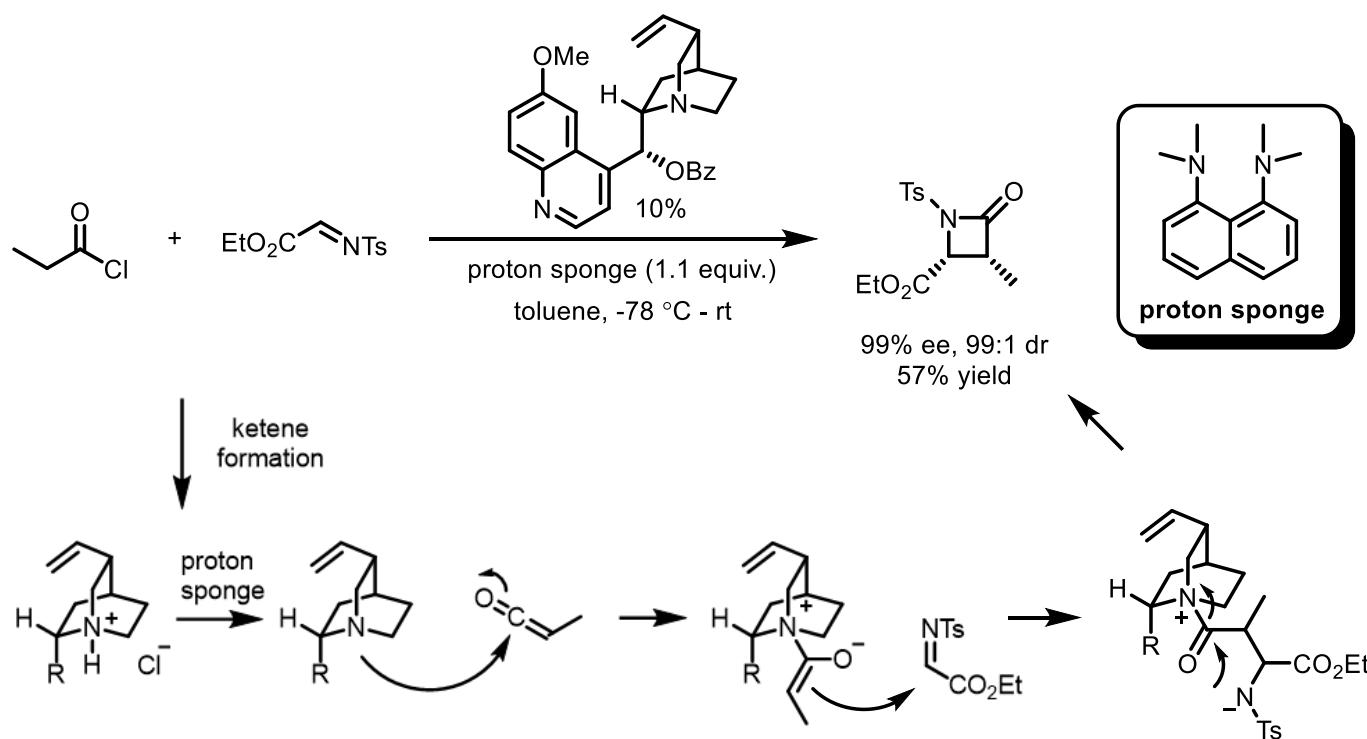


1. Provide the product and a mechanism for the named rearrangement.



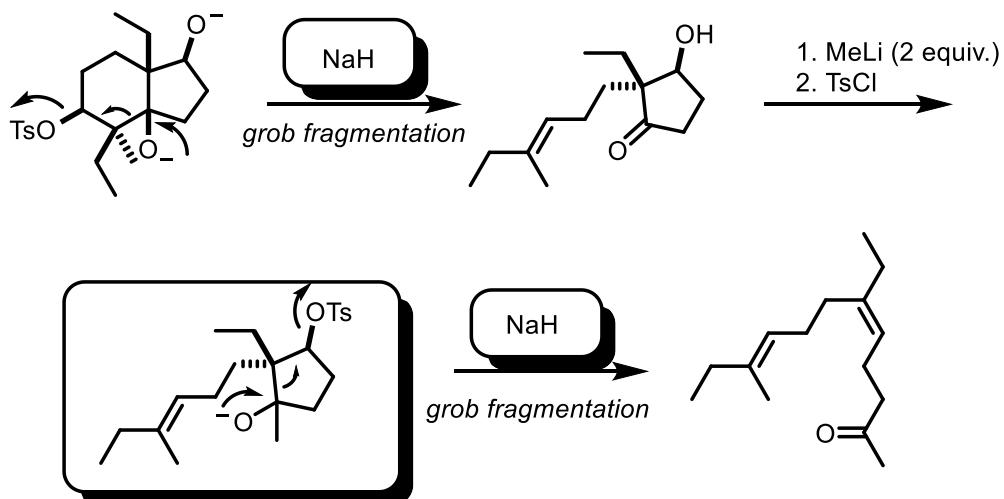
Clark, J.S. *Angew. Chem., Int. Ed.* **2016**, 55, 1–5

2. Provide a mechanism for the following transformation.



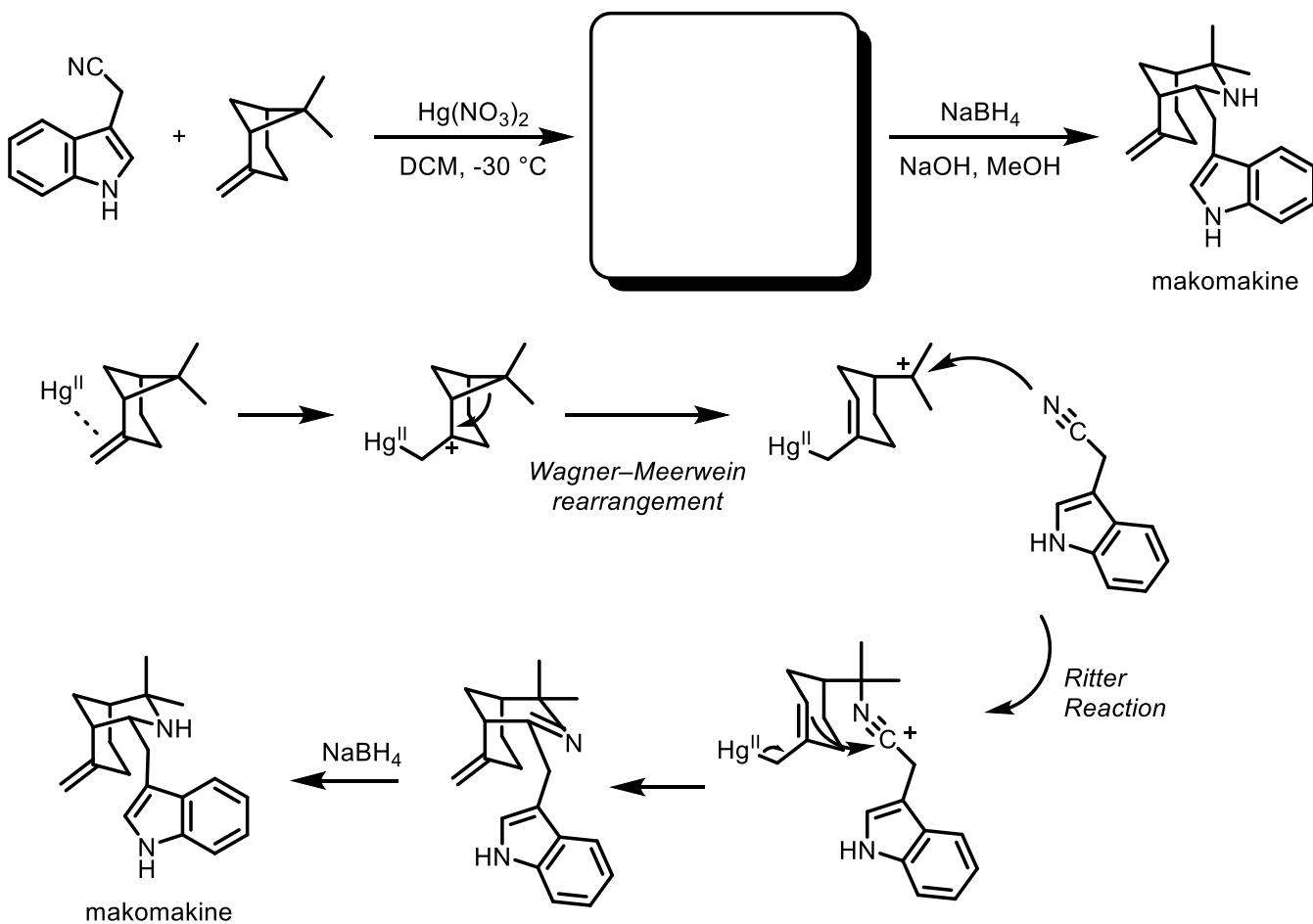
Lectka, T. *JACS*. **2000**, 122, 7831–7832

3. Provide the *simple* reagent needed for the following transformation. Propose a mechanism for the fragmentations.



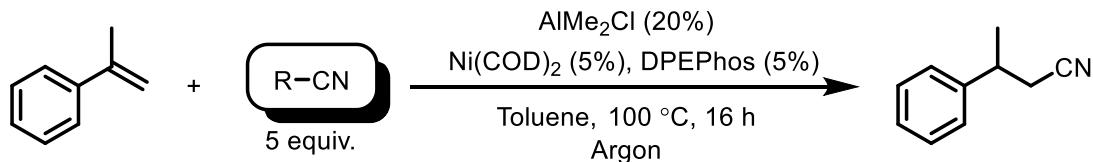
Edwards, J.A. *JACS*, **1968**, *90*, 6224–6225

4. Provide a mechanism for this concise synthesis of makomakine.

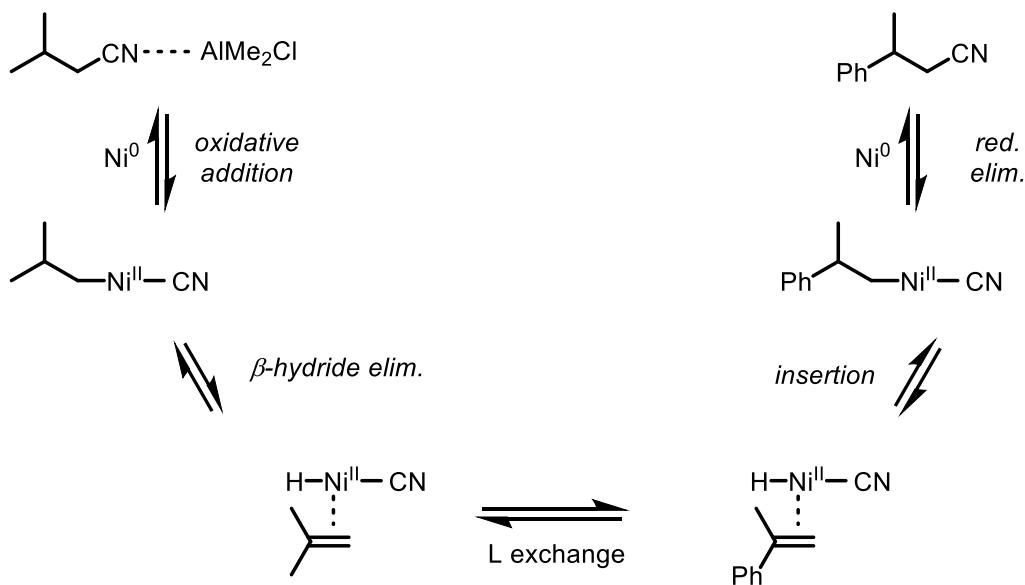


Stevens, R.V. *J. Chem. Soc., Chem. Commun.*, **1983**, 384-386

5. Propose a mechanism for the following transformation. Provide a rationale for the observed trend in yield.

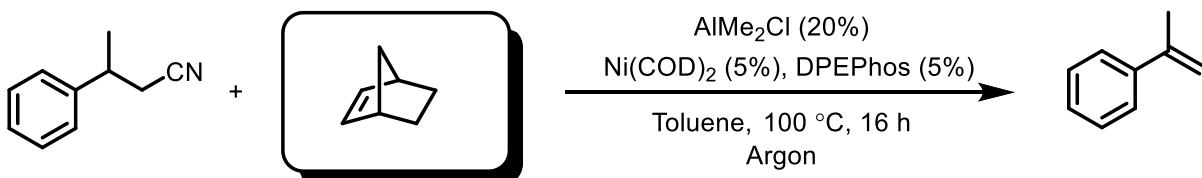


Nitrile:	$\text{CH}_3\text{CH}_2\text{CN}$	$\text{CH}_3\text{CH}(\text{CH}_3)\text{CN}$	$\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CN}$
Yield:	41	83	67



Yield determined by thermodynamics. Formation of disubstituted alkene more favorable to monosubstituted. Formation of volatile alkenes push equilibrium towards product.

How would you achieve the reverse reaction?



Use a strained alkene to drive the reaction. e.g. Norbornene or Norbornadiene.